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## Allyl bromide, 99%, stabilized

MSDS Specifications Applications Categories 3D model Infrared Molfile



## General

Product Name Allyl bromide  
3-Bromopropene

CAS RN 106-95-6

ACD Code MFCD00000244

Structure

Molecular Formula C3 H5 Br

Molecular weight 120.98

## Pack size

Catalog	Qty / UM	Price (USD)	
102900025	2.5 LT	289.40	<a href="#">Order</a> <a href="#">Check stock</a> Glass bottle
102900050	5 LT	486.20	<a href="#">Order</a> <a href="#">Check stock</a> Plastic bottle
102900100	10 LT	818.80	<a href="#">Order</a> <a href="#">Check stock</a> Plastic Jerry can
102902500	250 ML	46.20	<a href="#">Order</a> <a href="#">Check stock</a> Glass bottle
102905000	500 ML	74.20	<a href="#">Order</a> <a href="#">Check stock</a> Glass bottle

## Physical

Density (g/cm<sup>3</sup>) 1.39  
Refractive index 1.468 - 1.47  
Boiling Point (°C) 70 - 71  
Melting Point (°C) -119  
Flash Point (°C) -1

## Safety

## GHS Pictogram



GHS Signal Word Danger

## GHS H statement

H314: Causes severe skin burns and eye damage  
H331: Toxic if inhaled  
H301: Toxic if swallowed  
H400: Very toxic to aquatic life  
H225: Highly flammable liquid and vapor  
H340: May cause genetic defects  
H350: May cause cancer

## GHS P statement

P210: Keep away from heat/sparks/open flames/hot surfaces. - No smoking  
P303 + P361 + P353: IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water/shower  
P280: Wear protective gloves/protective clothing/eye protection/face protection  
P301 + P330 + P331: IF SWALLOWED: Rinse mouth. Do NOT induce vomiting  
P305 + P351 + P338: IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing  
P310: Immediately call a POISON CENTER or doctor/physician

## Categories

Functional Reagents &gt; Protection and Derivatization &gt; Protecting and Derivatizing Reagents &gt; Others

## Applications

Function	Transformation	Type	Caveat	Reference
Reagent	Alkylation	Allylation of carbonyl compounds	Using Zn in THF.	TL 1995, 36, 4885
Reagent	Alkylation	C-Alkylation	C-allylation is promoted with powdered potassium hydroxide alone or with tetra-n-butylammonium bromide in dioxane.	ZOR 1986, 22, 706
Reagent	Organometallic Chemistry	Wurtz coupling	With Mg in THF.	OS 1998, 76, 221

	Reagent	Protecting Group Chemistry	Protection of alcohols	Alcohols are allylated. This protecting group is removed by Pd (0) catalysis. JOC 1973, 38, 3224
	Reagent	Protecting Group Chemistry	Protection of carboxylic acids	Allyl esters from Cs salts reacting with allyl bromide. Int J. Pept Prot Res. 1985, 26, 493
Other				
Infrared	<a href="#">Show</a>			
Parameter	EINECS	203-446-6		
	Solubility	Solubility in water: insoluble Solubility in other solvents: miscible with alcohol, chloroform, ether, carbontetrachloride and carbon disulfide		
	Origin	synthetic		
	References: Literature	Lead-promoted allylation of carbonyl compounds with allyl bromide: S.Torii, et al; Chem. Lett. 9, 1611-14 (1986).		
	Reference: Beilstein	01,201		
	Reference: Fieser	13,219		
	Reference: Merck	15,279		
3D model	<a href="#">Show</a>			